A New Heteroannulation Method to 2-Cyclohexenone, Mediated by Phosphonate Auxiliaries. Synthesis of 4,5,6,7-Tetrahydrobenzothiazole Derivatives

E. Öhler

Institut für Organische Chemie, Universität Wien, Währinger Straße 38, A-1090 Wien, Austria

Summary. Dimethyl (1,2-epoxy-3-oxocyclohex-1-yl)phosphonate 7, easily available from 2-cyclohexen-1-one, affords the new (tetrahydrobenzothiazolyl)phosphonates 10 upon reaction with thiocarboxamides 8. This cyclocondensation proceeds with regioselective conjunction of the bisnucleophile 8 to carbonyl group and adjacent oxirane carbon of the epoxide precursor. On treatment with alkali the 5,6-dihydro-7(4H)benzothiazolones 12 are obtained from compounds 10, or from the intermediately formed bicyclic dihydroxy derivatives 9. Reaction of the corresponding cyclopentylphosphonate 14 with thiobenzamide (8b) yields the cyclopentathiazole derivative 15 as a single isomer, the dihydro-thiazole moiety being now annulated regio- and *cis*-stereoselectively to both oxirane carbons of the epoxyphosphonate.

Keywords. (1,2-Epoxy-3-oxo-cyclohex-1-yl)phosphonate; 5,6-Dihydro-7(4*H*)benzothiazolones, 2-substituted; Heteroannulation with thiocarboxamides; (4,5,6,7-Tetrahydrobenzothiazol-7-yl)phosphonates; Thiocarboxamides.

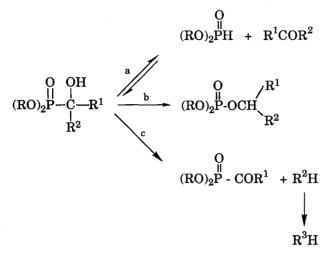
Eine neue, durch Phosphonat-Synthone gesteuerte Methode zur Heteroanellierung an 2-Cyclohexenon. Synthese von 4,5,6,7-Tetrahydrobenzthiazol-Derivaten

Zusammenfassung. Umsetzung des aus 2-Cyclohexen-1-on leicht zugänglichen (1,2-Epoxy-3-oxocyclohex-1-yl)phosphonats 7 mit den Thioamiden 8 liefert die neuen (Tetrahydrobenzthiazolyl)phosphonate 10. Dabei wird der Thiazolring regioselektiv an Carbonylgruppe und benachbarten Oxirankohlenstoff des Epoxyphosphonats anelliert. Mit Alkali werden sowohl die α -Hydroxyphosphonate 10 als auch die intermediär gebildeten Dihydroxyverbindungen 9 zu den 5,6-Dihydro-7(4H)benzthiazolonen 12 gespalten. Im Gegensatz dazu reagiert das Cyclopentylphosphonat 14 mit Thiobenzamid 8b zu einem Cyclopentathiazol-Derivat 15, bei dem der Dihydrothiazolring regiound *cis*-stereoselektiv mit den beiden früheren Oxiran-Kohlenstoffen verknüpft ist.

Introduction

The behaviour of α -hydroxyphosphonates to basic reagents depends on the substituents R¹ and R² (Scheme 1). Thus strong bases in aqueous solutions cause the breakdown of the P–C-bond with formation of carbonyl compounds and dialkyl-

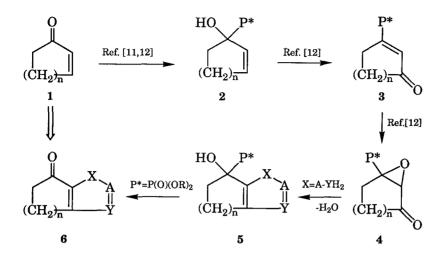
phosphites, when \mathbb{R}^1 , $\mathbb{R}^2 = \mathbb{H}$, alkyl [1] (way a). If one of the substituents is an electron attracting group able to stabilize a carbanionic intermediate, rearrangement to a phosphate derivative will occur [2] (way b). Occasionally, cleavage of the β -C-C-bond has been observed, particularly if this step was accompanied by subsequent aromatization ($\mathbb{R}^2\mathbb{H} \to \mathbb{R}^3\mathbb{H}$) of the leaving group (way c) [3].



Scheme 1

During the last years acylic (1,2-epoxy-3-oxoalkyl)phosphonates, and the homologous 2,3-epoxy-4-oxo derivatives have been shown in our group to be versatile intermediates for the synthesis of phosphorus and non-phosphorus bearing heterocycles [4–8]. The 1.2-biselectrophilic behaviour exhibited by these compounds was manifested by a series of regioselective cyclocondensation reactions with their polarity complements, 1,3-bidentate nucleophiles $X = A - YH_2$ (Scheme 2): Thiocarboxamides, activated a-picolines, and 2-aminopyri(mi)dines, reagents for the well established heterocyclization reactions with α -haloketones leading to thiazoles [9]. indolizines [10], and imidazo[1,2-a]pyri(mi)dines [10], respectively, have been reacted analogously with the acyclic epoxyphosphonates 4 to yield (α -hydroxy- α heteroaryl)-[4,5], or $(\beta$ -hydroxy- β -heteroaryl)alkylphosphonates [6] (monocyclic compounds of the general type 5). The hydroxy group adjacent to the heterocycle, generated by regioselective epoxide opening by the more nucleophilic part of the bisnucleophile, can be used for various further transformations [4-6, 8]. Especially the heteroaryl-substituted α -hydroxyphosphonates, upon treatment with alkali have been transformed to a series of acylsubstituted heterocycles according to way a, Scheme 1 ($\mathbf{R}^1 = \mathbf{H}$, alkyl, $\mathbf{R}^2 =$ thiazol-5-yl, indolizin-3-yl, imidazo[1,2-a]pyri(mi)din-3-yl) [5].

Recently we found a convenient route to the new (1,2-epoxy-3-oxo-cycloalkyl) phosphonates and the homologous cycloalkylmethyl derivatives 4 [P* = P(O)(OR)₂, and CH₂P(O)(OR)₂, respectively]. This synthesis (regioselective 1,2-addition of the required phosphorus nucleophiles to cycloalkenones 1 [11, 12], followed by oxidative 1,3-transposition of the tertiary allylic hydroxy group with chromium(VI) reagents,



 $P^* = P(O)(OR)_2, CH_2P(O)(OR)_2$



and subsequent epoxidation [12], Scheme 2) centered around the expectation to develop a new heteroannulation procedure to the double bond of cycloalkenones 1 following the pathway presented in Scheme 2, the key step being the regioselective conjunction of the epoxyphosphonates 4 with bisnucleophiles of the general type $X = A-YH_2$. In the next step elimination of the auxiliary dialkyl phosphite from the heterocyclic α -hydroxyphosphonate 5 [P* = P(O)(OR)₂] would regenerate the carbonyl group of the cycloalkenone precursor 1.

This paper presents the first results gained upon cyclocondensation of the epoxyphosphonates 7 (and 14) with a series of thiocarboxamides 8.

Results and Discussion

Reaction of the cyclohexyl phosphonate 7 with thiocarboxamides 8 in boiling ethanol afforded satisfactory yields of the desired bicyclic annulation products 10, using the thioamides 8b, 8c, and 8f (R = Ph, 3-pyridyl, and CO₂Et, respectively). The derivative 10e, which was found to decompose in boiling ethanol, was also obtained in good yield, when 7 was reacted with thiourea at room temperature (Table 1). Upon condensation of 7 with the 4-pyridyl derivative 8d, the α -hydroxyphosphonate 10d was formed sluggishly (yield 29%), partial deprotection of the phosphonate ester, and formation of ketone 12d being observed by TLC monitoring. However, an intermediate cyclization product was isolated after a short reaction time and before decomposition was apparent, whose ¹³C-NMR data suggested the dihydrothiazole structure 9d (1:1 mixture of two isomeric compounds, yield 66%), depicted in Scheme 3. This dihydroxy compound, after the usual treatment with alkali (0.5 N NaOH, room temp.) afforded the benzothiazolone derivative 12d in

Thioamide	hioamide Reaction time (h)	Temp. (°C)	Product(s)	Yield ^a (%)	M.p. (°C)	Cleavage method ^b	Reaction time (min)	Product	Yield ^a (%)	M.p. (°C)
Sa	1.5	80				B	30	12a	09	2
8b	6	80	10b	88	139–142	A	20	12b	95	114-115
86	12.5	80	10c	55	131-134	Α	30	12c	67	114-115
8d	+	80	9dd	99	ł	Α	15	12d	95	155-157
8d	1	80	.	ļ	İ	В	15	12d	74	t
8d	22	80	10d	29	148 dec.	A	15	12d	98	I
8e	7d	20	10e ^e	69	$158-160^{\circ}$	A	20	12e°	95	272°
			13e [°]	9	5760°					
%	7d	20	I	I	i	B	20	12e ^e	83	1
8f	24	80	10f	67	138 - 140	C	100	12f	78	85-87

^b Method A: crystalline 10 (9) + 0.5 N NaOH, room temp; Method B (one pot procedure): 1, 7+8 (ethanol), 2, 0.5 N NaOH, room temp; Method C: satd aq. NaHCO₃, room temp. • Oil, b.p. 80-85°C (oven temp. at Kugelrohr distillation) at 0.01 Torr, Ref. [14] 85-87°C/0.2 Torr

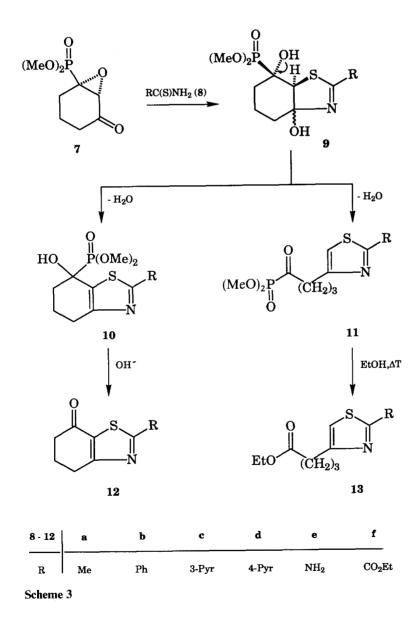
^d 1:1 - Mixture of isomers

N-Acetyl derivative

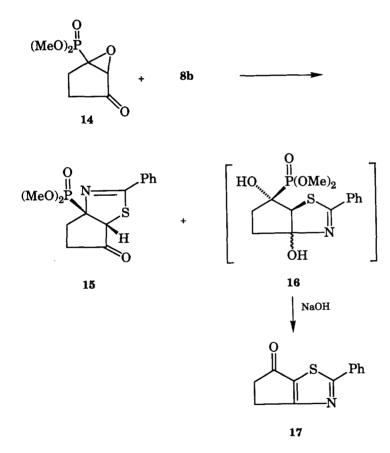
very good yield. Similar results have been obtained performing cyclization of 7 with 8d and cleavage of the intermediate 9d in an one pot procedure (Table 1).

Thioacetamide 8a, which with acyclic epoxyphosphonates has been found to give only poor results [5], yielded also the carbonyl derivative 12a in 60% overall yield upon application of the one pot procedure.

The formation of the open chain thiazole derivative 13e, which was isolated as byproduct reacting epoxide 7 with thiourea under various reaction conditions, can also be explained assuming a dihydrothiazole intermediate 9e: Retroaldol reaction at C-7a and subsequent dehydration, which parallels way c, Scheme 1, would lead to the thiazolyl-substituted acylphosphonate 11e, which then is solvolyzed to generate the ethylester derivative 13e observed.



Under analogous reaction conditions, condensation of dimethyl (1,2-epoxy-3oxo-cyclopent-1-yl)phosphonate 14 with thiobenzamide 8b afforded no (or only traces) of an α -hydroxyphosphonate. Instead, an isomeric compound was isolated with high yield, whose spectral data suggest the bicyclic structure 15 shown in Scheme 4: The IR spectrum exhibits no OH-absorption, but a strong absorption at 1752 cm⁻¹, which is typical for cyclopentanones. The base peak of the mass spectrum was found to be 216 (M-109), as compared with the spectra of compounds 10, which uniformely exhibited no molecular peaks, but an intensive fragmentation, caused by the loss of dimethylphosphite (M-110). Further evidence for structure and *cis*-stereochemistry of compound 15 is found in the ¹H-NMR (doublet of a CH-signal at $\delta = 4.45$ ppm with a vicinal ³¹P-C-C-H coupling, ³J_{PH} = 16.4 Hz, which suggests the *cis*-relation) and also in the ¹³C-NMR-spectrum (CO-signal at $\delta = 211.91$ ppm with a high vicinal ³¹P-C-CC-CO coupling, ³J_{PC} = 15.2 Hz, which is also in accordance with the geometry given in the scheme) [13].



Scheme 4

The formation of compound 15, which regioselectively integrates the both oxirane carbons of the precursor 14 in the dihydrothiazole moiety with the carbonyl group being unchanged, parallels the annulation type previously observed in our laboratory, upon the condensation of acyclic epoxyphosphonates with some 2-mercaptoazoles [7]. There, the generation of acyl-substituted dihydrothiazoles

has been explained by regioselective oxirane opening by the sulfur group, subsequent dehydration, and intramolecular Michael addition (5-endo trig cyclization) of the nitrogen part of the bisnucleophile to the α -enone formed intermediately.

Experimental Part

Melting points were determined with a Kofler apparatus and are uncorrected. IR-spectra (CH_2Cl_2) were recorded on a Perkin-Elmer 377 Infrared spectrophotometer. ¹H- and *J*-modulated ¹³C-NMR spectra were recorded on a Bruker WM 250 spectrometer. TLC was performed on Merck silica gel 60 F_{254} plates with visualization by UV light (254 nm) and/or by iodine vapour. Flash-chromatography was performed on glass columns packed with Merck silica gel 60 (230–400 mesh). The epoxy phosphonates 7 and 14 were prepared according to the literature protocol [12].

Dimethyl [4,5,6,7,3a,7a-Hexahydro-3a,7-dihydroxy-2-(4-pyridyl)benzothiazol-7-yl]phosphonate (9d)

A solution of 7 and 8d (each 1.0 mmol) in dry EtOH (5 ml) was refluxed, the reaction being monitored by TLC (EtOAc/MeOH, 5:1) to avoid the formation of 10d. After 1 h the solution, which was shown to contain unreacted starting materials and a substance slightly more polar than 10d (9d: $R_f = 0.26$, TLC-visualization at 254 nm; 10d: $R_f = 0.29$, visualization at 254 and 366 nm) was evaporated. The residue was subjected to flash-chromatography (EtOAc/MeOH, 5:1) to yield subsequently a mixture of 7 and 8d ($R_f \approx 0.75$, 160 mg), and 9a (182 mg), colorless crystals, not soluble in CH₂Cl₂ and CHCl₃. The above procedure was repeated with unreacted material to give further 56 mg of 9a. Total yield: 238 mg (66%). ¹³C-NMR (*DMSO-d*₆): $\delta = 14.17$ (³ $J_{PC} = 10.0$), 15.06 (³ $J_{PC} = 11.5$ Hz) [C-5], 26.49, 28.58, 31.78, 33.75 (4s, C-4, C-6), 53.40, 53.52, 53.66, 53.79 (4d, $J_{PC} = 8.2, 8.2, 8.3,$ and 6.8 Hz, POCH₃), 57.63 (s), 61.29 (d, ² $J_{PC} = 5.1$ Hz) [C-7a], 71.61, 71.73 (2d, ¹ $J_{PC} = 167.3$, and 172.1 Hz, C-7), 104.94, 105.20 (2d, ³ $J_{PC} = 13.7$, and 12.0 Hz, C-3a), 121.09, 121.58 (C-3', C-5'), 139.77, 140.01 (C-4'), 150.41, 150.52 (C-2', C-6'), 161.00, 165.43 (C-3) [diastereomeric ratio $\approx 1:1$]. C₁₄H₁₉N₂O₅PS (358.4). Calc. C 46.91, H 5.35, N 7.82, S 8.95; found C 46.72, H 5.43, N 7.71, S 8.60.

Synthesis of 2-Substituted Dimethyl [(4,5,6,7-Tetrahydro-7-hydroxy)benzothiazol-7-yl]phosphonates **10** (General Procedure)

A solution of epoxyphosphonate 7 (220 mg, 1.0 mmol) and thioamide 8 (1.0 mmol) in dry EtOH (5 ml) is refluxed (compounds 8b, 8c, 8d, 8f) or reacted at room temp. (compound 8e) under Ar, the reaction being monitored by TLC. After complete consumption of 7 and the polar intermediates primarily formed (reaction time given in Table 1), the solution is evaporated and the residue purified by flash-chromatography on silica gel (recrystallization of compounds 10 should be avoided, as upon heating, tertiary α -hydroxyphosphonates are partially cleaved to the corresponding carbonyl derivatives and dialkylphosphite).

Dimethyl [(4,5,6,7-Tetrahydro-7-hydroxy-2-phenyl)benzothiazol-7-yl]phosphonate (10b)

Flash-chromatography (CH₂Cl₂/acetone, 4:1, $R_f = 0.18$), yield 300 mg (88%), colorless crystals, m.p. 139–142 °C. IR(CH₂Cl₂): v = 3560 cm⁻¹, 3270 (OH), 1235 (P=O), 1060, 1038 (P–OC). ¹H-NMR (CDCl₃): $\delta = 1.95-2.30$ (m, 4H, H-5, H-6), 2.70–3.00 (m, 2H, H-4), 3.78, 3.82 (2d, $J_{\rm HP} = 10.0$ Hz, each 3H, POCH₃), 4.08 (br d, $J_{\rm HP} \approx 5.5$ Hz, 1H, D₂O-exchange, OH), 7.40 (m, 3H_{arom}), 7.90 (m, 2H_{arom}). C₁₅H₁₈NO₄PS (339.4). Calc. C 53.08, 5.36, N 4.12, S 9.45; found C 53.03, H 5.37, N 4.16, S 9.49.

Dimethyl {[4,5,6,7-Tetrahydro-7-hydroxy-2-(3-pyridyl)]benzothiazol-7-yl}phosphonate (10c)

Flash-chromatography (EtOAc/MeOH, 19:1, $R_f = 0.21$), yield 188 mg (55%), m.p. 131–134 °C. ¹H-NMR (CDCl₃): $\delta = 1.96-2.37$ (m, 4H, H-5, H-6), 2.70–3.03 (m, 2H, H-4), 3.80, 3.81 (2d, J = 10.0 Hz, each 3H, POCH₃), ≈ 4.75 (br s, 1H, D₂O-exchange, OH), 7.34 (dd, 1H, H-5'), 8.15 (td, 1H, H-4'), 8.59 (dd, 1H, H-6'), 9.07 (d, 1H, H-2') [$J_{4',5'} \approx 8.1$, $J_{5',6'} \approx 5.0$, $J_{4',6'} \approx J_{2',4'} \approx 1.5$ Hz]. C₁₄H₁₇N₂O₄PS (340.4). Calc. C 49.40, H 5.04, N 8.23, S 9.42; found C 49.48, H 5.09, N 8.11, S 8.97.

Dimethyl {[4,5,6,7-Tetrahydro-7-hydroxy-2-(4-pyridyl)]benzothiazol-7-yl}phosphonate (10d)

Flash-chromatography (EtOAc/MeOH, 5:1, $R_f = 0.29$); yield 103 mg (29%), m.p. 148 °C (dec.) from EtOAc. Compound 10d is not stable in solution, ketone 12d and a polar decomposition product being detected by TLC after a few hours. ¹H-NMR (CDCl₃): $\delta = 1.91-2.35$ (m, 4H, H-5, H-6), 2.67-3.00 (m, 2H, H-4), 3.75, 3.76 (2d, $J_{HP} = 10.3$ and 10.4 Hz, each 3H, POCH₃), ≈ 5.1 (br s, 1H, D₂O-exchange, OH), 7.68 (d, 2H, H-3', H-5'), 8.54 (d, 2H, H-2', H-6'). C₁₄H₁₇N₂O₄PS (340.4). Calc. C 49.40, H 5.04, N 8.23, S 9.42; found C 49.17, H 5.01, N 8.21, S 9.39.

N-[(7-Dimethoxyphosphoryl-4,5,6,7-tetrahydro-7-hydroxy)benzothiazol-2-yl]acetamide (Ac-10e)

A solution of 7 (2.20 g, 10.0 mmol) and thiourea (760 mg, 10.0 mmol) in dry EtOH (50 ml) was stirred under argon at room temp. After 3 h crystals started to deposit. Stirring was continued for 7 days, then **10e** (860 mg) was filtered off. The mother liquor was evaporated, the dark residue treated with EtOAc to crystallize another portion (630 mg) of **10e**. The residue obtained from the filtrate was subjected to flash-chromatography (EtOAc/MeOH, 19:1) to isolate subsequently **13e** ($R_f = 0.75$, 138 mg, 6%), and 415 mg **10e** (total yield: 1.905 g, 69%, $R_f = 0.24$) as yellowish crystals.

For analysis, both compounds have been reacted with Ac_2O at room temp. to give the corresponding N-acetyl derivatives.

Ac-10e: M.p. 158–160 °C (MeOH/EtOAc). ¹H-NMR (*DMSO-d*₆): $\delta = 1.77-2.11$ (m, 4H, 2xCH₂), 2.11 (s, 3H, COCH₃), 2.42–2.69 (m, 2H, CH₂), 3.67, 3.71 (2d, J = 10.0 Hz, each 3H, POCH₃), 6.07 (d, ³ $J_{HP} \approx 13$ Hz, 1H, exchange with D₂O, OH), ≈ 12.0 (br s, 1 H, exchange with D₂O, NH). ¹³C-NMR (*DMSO-d*₆): $\delta = 17.44$ (³ $J_{PC} = 6.8$ Hz, C-5), 21.89 (COCH₃), 25.69, 32.70 (C-4, C-6), 52.81, 52.97 (² $J_{PC} = 7.4$, and 7.5 Hz, POCH₃), 68.96 (¹ $J_{PC} = 176.9$ H, PC), 121.26 (C-7a), 147.43 (³ $J_{PC} = 10.0$ Hz, C-3a), 157.53 (C-2), 167.65 (NHCO). MS (70 eV): m/z (⁶₀) = 320 (1) [*M*⁺], 302 (50) [*M*-H₂O], 260 (100) [*M*-H₂O-CH₂CO]. C₁₁H₁₇N₂O₅PS (320.3). Calc. C 41.24, H 5.36, N 8.75, S 10.01; found C 41.69, H 5.48, N 8.28, S 9.89.

N-[4-(3-Ethoxycarbonyl)propyl-thiazol-2-yl]acetamid (Ac-13e)

M.p. 57–60 °C (ether/petroleum ether, b.p. 40 °C). ¹H-NMR (CDCl₃): $\delta = 1.26$ (t, 3H, OCH₂CH₃), 2.00 (quint, CH₂CH₂CH₂), 2.24 (s, 3H, COCH₃), 2.35, 2.70 (2t, each 2H, CH₂-CH₂-CH₂), 4.14 (q, 2H, CO₂CH₂CH₃), 6.58 (s, 1H, H-5). ¹³C-NMR (CDCl₃): $\delta = 14.25$ (CO₂CH₂CH₃), 23.18 (COCH₃), 24.12 (CH₂-CH₂-CH₂), 30.47, 33.53 (CH₂-CH₂-CH₂), 60.39 (CO₂CH₂), 108.13 (C-5), 149.96 (C-4), 158.26 (C-2), 167.80 (NHCO), 173.34 (CO₂). MS (70 eV): m/z (%) = 256 [M^+], 211 (15) [M-OEt], 169(85), 127(100). C₁₁H₁₆N₂O₃S (256.4). Calc. C 51.52, H 6.30, N 10.93, S 12.51; found C 51.28, H 6.28, N 11.07, S 12.63.

Ethyl [7-Dimethoxyphosphoryl-4,5,6,7-tetrahydro-7-hydroxy-benzothiazol-2-yl]carboxylate (10f)

Flash-chromatography (EtOAc/MeOH, 19:1, $R_f = 0.56$); yield 225 mg (67%), colorless crystals, m.p. 138–140 °C (Et₂O). ¹H-NMR (CDCl₃): $\delta = 1.40$ (t, J = 7.0 Hz, 3H, CO₂CH₂CH₃), 1.92–2.38 (m, 4H), 2.71–3.08 (m, 2H) [H-4, H-5, H-6], 3.75, 3.82 (2d, $J_{HP} = 10.4$ Hz, each 3H, POCH₃), 4.45 (q, J = 7.0 Hz, CO₂CH₂CH₃), 4.80 (d, J = 3.7 Hz, 1H, D₂O-exchange, OH). C₁₂H₁₈NO₆PS (335.4). Calc. C 42.98, H 5.42, N 4.18, S 9.56; found C 43.01, H 5.37, N 4.14, S 9.37.

Synthesis of 5,6-Dihydro-7(4H)benzothiazolones 12

a) From α -Hydroxyphosphonates 10 (General Procedure): A suspension of 10 (1.0 mmol) in 10 ml 0.5 N aq. NaOH (compounds 10a-10e), or 10 ml satd. aq. NaHCO₃ (10f) is stirred at room temp. for the time given in Table 1 (work up is given below).

b) From 7 and 8 (One Pot Procedure): A solution of 7 and 8 (each 1.0 mmol) in dry EtOH (5 ml) is reacted under the conditions given in Table 1, and then evaporated. To the dry residue 0.5 N NaOH (10 ml) is added with stirring at room temp. (work up is given below).

5,6-Dihydro-2-methyl-7(4H)benzothiazolone (12a) (One Pot Procedure)

The dark aqueous solution was extracted with CH_2Cl_2 , the dry (Na₂SO₄) extracts were evaporated and the residue purified by Kugelrohr distillation (oven temp. 80–85 °C, 0.01 Torr, Ref. [14] b.p. 85–87 °C/0.2 Torr) to yield **12a** (101 mg, 60%) as a colorless oil, $R_f = 0.40$ (CH₂Cl₂/EtOAc, 9:1). – ¹H-NMR (CDCl₃): $\delta = 2.09$ (quint, J = 6.0 Hz, 2H, H-5), 2.49 (br t, $J \approx 6$ Hz, 2H, H-4), 2.62 (s, 3H, CH₃), 2.90 (t, J = 6.0 Hz, 2H, H-6).

5,6-Dihydro-2-phenyl-7(4H)benzothiazolone (12b) from 10b

The reaction mixture was cooled to 5 °C, **12b** was filtered off, washed with cold H₂O, and dried at 0.01 Torr. Yield 217 mg (95%), colorless crystals, m.p. 114–115 °C (ether, petroleum ether b.p. 40 °C), Ref. [14] m.p. 121–123 °C. IR (CH₂Cl₂): $v = 1668 \text{ cm}^{-1}$ (CO). ¹H-NMR (CDCl₃): $\delta = 2.24$ (quint, J = 6.0 Hz, 2H, H-5), 2.64 (t, J = 6.0 Hz, 2H, H-4), 3.09 (t, J = 6.0 Hz, 2H, H-6), 7.45 (m, 3H_{arom}), 7.96 (m, 2H_{arom}). MS (70 eV): m/z (%) = 229 (26) [M^+], 201 (40) [M–CO], 70 (100). C₁₃H₁₁NOS (229.3). Calc. C 68.09, H 4.85, N 6.11, S 13.98; found C 67.85, H 4.86, N 6.10, S 13.81.

5,6-Dihydro-2-(3-pyridyl)-7(4H)benzothiazolone (12c) from 10c

The aqueous reaction mixture was extracted with CH₂Cl₂. The residue, obtained from the dried (Na₂SO₄) extracts was purified by flash-chromatography (CH₂Cl₂/ EtOAc, 3:7) to yield **12c** (223 mg, 97%) as colorless crystals m.p. 114–115 °C (Et₂O). IR (CH₂Cl₂): $v = 1670 \text{ cm}^{-1}$ (CO). ¹H-NMR (CDCl₃): $\delta = 2.24$ (quint, J = 6.0 Hz, 2H, H-5), 2.63 (t, J = 6.0 Hz, 2H, H-4), 3.10 (t, J = 6.0 Hz, 2H, H-6), 7.38 (dd, 1H, H-5'), 8.23 (td, 1H, H-4'), 8.68 (br dd, 1H, H-6'), 9.18 (br d, 1H, H-2') [$J_{4',5'} = 8.0$, $J_{5',6'} = 5.0$, $J_{2',4'} \approx J_{4',6'} \approx 1.5 \text{ Hz}$]. C₁₂H₁₀N₂OS (230.3). Calc. C 62.58, H 4.39, N 12.17, S 13.92; found C 62.83, H 4.25, N 12.12, S 13.70.

5,6-Dihydro-2-(4-pyridyl)-7(4H)benzothiazolone (12d)

a) From 10d: Work up analogously to 12c afforded 12d (225 mg, 98%) as colorless crystals, m.p. 155–157 °C (Et₂O). ¹H-NMR (CDCl₃): $\delta = 2.26$ (quint, J = 6.0 Hz, 2H, H-5), 2.66 (t, J = 6.0 Hz, 2H, H-4), 3.07 (t, J = 6.0, 2H, H-6), 7.80 (d, J = 5.5 Hz, 2H, H-3', H-5'), 8.74 (d, J = 5.5 Hz, 2H, H-2', H-6'). MS (70 eV): m/z (%) = 230 (76) [M^+], 202 (66) [M-CO], 70 (100). C₁₂H₁₀N₂OS (230.3). Calc. C 62.58, H 4.39, N 12.17, S 13.92; found C 62.44, H 4.39, N 11.93, S 13.90.

b) From 7 and 8d (One Pot Procedure): 12d was extracted with CH_2Cl_2 and purified by flash-chromatography ($CH_2Cl_2/EtOAc$, 3:7). Yield 171 mg (74%).

c) From 9d: 9d (100 mg, 0.28 mmol) was reacted with 0.5 N NaOH according to the protocol given for the conversion of 10d to 12d. Yield 61 mg (95%).

N-[(4,5,6,7-Tetrahydro-7-oxo)benzothiazol-2-yl]acetamide (Ac-12e) (One Pot Procedure)

A solution of 7 and 8e (each 1.0 mmol) in dry EtOH (5 ml) was reacted at room temp. as given for the synthesis of 10e and then evaporated. To the residue was added 0.5 N aq. NaOH (10 ml) with stirring at room temp. After 30 min the reaction mixture was brought to pH = 7 with 1 N HCl and then evaporated to dryness. The residue was purified by flash-chromatography (EtOAc/MeOH, 19:1, $R_f = 0.47$) to yield 141 mg (83%) 12e, yellowish crystals m.p. 278–280 °C (Ref. [14] m.p. 280–282 °C). Acetylation with Ac₂O at room temp. afforded, after usual work up Ac-12e: m.p. 272 °C (MeOH). ¹H-NMR (DMSO-d₆): $\delta = 2.09$ (m, 2H, H-5), 2.18 (s, 3H, COCH₃), 2.50 (t, 2H, H-4), 2.85 (t, 2H, H-6),

12.6 (br s, 1H, D₂O-exchange, NH). ¹³C-NMR: δ = 22.58 (COCH₃ + C-5), 26.32 (C-4), 37.15 (C-6), 123.22 (C-7a), 162.71, 163.97 (C-2, C-3a), 169.17 (NHCO), 191.85 (CO). C₉H₁₀N₂O₂S (210.3). Calc. C 51.41, H 4.80, N 15.25; found C 50.73, H 4.69, N 14.53.

Ethyl (4,5,6,7-Tetrahydro-7-oxobenzothiazol-2-yl)carboxylate (12f) from 10f

Purification by flash-chromatography (CH₂Cl₂/EtOAc, 3:7, $R_f = 0.67$), yield 175 mg (78%), yellowish crystals, m.p. 85–87 °C (ether/petroleum ether, b.p. 40 °C). ¹H-NMR (CDCl₃): $\delta = 1.40$ (t, J = 7.0 Hz, 3H, CO₂CH₂CH₃), 2.20 (quint, J = 6.0 Hz, 2H, H-5), 2.61 (t, J = 6.0 Hz, 2H, H-4), 3.10 (t, J = 6.0 Hz, 2H, H-6), 4.45 (q, J = 7.0 Hz, 2H, CO₂CH₂). MS (70 eV): m/z (%) = 225 (18) [M^+], 197 (19) [M-CO], 153 (100). C₁₀H₁₁NO₃S (225.3). Calc. C 53.31, H 4.93, N 6.22, S 14.23; found C 53.88, H 5.12, N 6.10, S 14.20.

Dimethyl [3a,5,6,6a-Tetrahydro-6-oxo-2-phenyl-3a(4H)cyclopentathiazolyl]phosphonate 15

A solution of 14 (206 mg) and **8b** (137 mg) (each 1.0 mmol) in dry EtOH (5 ml) was refluxed, the reaction being monitored by TLC (CH₂Cl₂/acetone, 4:1). The starting thioamide ($R_f \approx 0.75$) was consumed after 1–2 h, the primary reaction products ($R_f \leq 0.1$) being slowly converted to a less polar compound ($R_f = 0.38$). After 15 h the dark solution was evaporated. The residue, on flash-chromatography over silica gel (20 g) afforded 267 mg (82%) 15, colorless crystals, m.p. 80–81 °C (ether/petroleum ether, b.p. 40 °C). IR (CH₂Cl₂): v = 1753 cm⁻¹ (CO), 1248 (P=O), 1058, 1030 (P–OC). MS (70 eV): m/z (%) = 325 (2.4) [M^+], 216 (100) [M–P(O)(OMe)₂]. ¹H-NMR (CDCl₃): $\delta = 2.30-2.86$ (m, 4H, 2CH₂), 3.80, 3.84 (2d, J = 10.0 Hz, each 3H, POCH₃), 4.45 (d, ³ $J_{HP} = 16.4$ Hz, CH), 7.34–7.52 (m, 3H_{arom}), 7.82 (m_c 2H_{arom}). ¹³C-NMR (CDCl₃): $\delta = 30.37$ (s), 34.75 ($J_{PC} = 5.1$ Hz) [CH₂], 53.79, 54.00 ($^2J_{PC} = 6.8$, and 7.2 Hz, POCH₃), 56.78 ($^2J_{PC} = 4.3$ Hz, CH), 88.15 ($^1J_{PC} = 168.5$ Hz, P–C), 128.47 (4CH_{arom}), 131.69 ($^4J_{PC} = 2.6$ Hz, *i*-C_{arom}), 131.90 (*p*-CH_{arom}), 169.12 ($^3J_{PC} = 14.3$ Hz, C-2), 211.91 ($^3J_{PC} = 15.2$ Hz, CO). C₁₄H₁₆NO₄PS (325.4). Calc. C 51.86, H 4.97, N 4.31, S 9.85; found C 51.58, H 4.92, N 4.33, S 9.90.

5,6-Dihydro-2-phenyl-6(4H)cycopentathiazolone 17

A small amount of polar reaction products was collected by further elution (EtOAc/MeOH, 5:1) of the column after the isolation of **15**. This was treated with 0.5 N NaOH (2 ml) for 15 min. Then the solution was extracted with CH₂Cl₂. Flash-chromatography of the dried (Na₂SO₄) solution (CH₂Cl₂/EtOAc, 1:1) afforded **17** (25 mg, 12%, colorless crystals with $R_f = 0.80$), m.p. 150–151 °C (ether/petroleum ether, b.p. 40 °C). IR (CH₂Cl₂): v = 1700 cm⁻¹, 1378, 1298. MS (70 eV): m/z (%) = 215 (55) [M^+], 84 (100). ¹H-NMR (CDCl₃): $\delta = 3.02$, 3.19 (2 m_c, each 2H, 2CH₂), 7.48 (m, 3H_{arom}), 8.00 (m_c, 2H_{arom}). C₁₂H₉NOS (215.3). Calc. C 66.95, H 4.22, N 6.59, S 14.89; found C 67.11, H 4.37, N 6.46, S 14.14.

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